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The fifth edition of this widely acclaimed work has been reissued as part of the Oxford Classic Texts series. The book includes a clear exposition of general topics concerning the structures of solids, and a systematic description of the structural chemistry of elements and their compounds. The book is divided into two parts.

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Structural Inorganic Chemistry. Alexander Frank Wells. Oxford  
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The First Edition of this book, which appeared in 2013, serves as a problem text for Part I (Fundamentals of Chemical Bonding) and Part II (Symmetry in Chemistry) of the book Advanced Structural Inorganic Chemistry published by Oxford University Press in 2008. A Chinese edition was published by Peking University Press in August in the same year.

### **Problems in Structural Inorganic Chemistry - Oxford ...**

Structural Inorganic Chemistry A. F. WELLS FIFTH EDITION \ CLARENDON PRESS • OXFORD . Contents v ABBREVIATIONS xxix PART I. INTRODUCTION 3 The importance of the solid State 3 Structural formulae of inorganic Compounds 11 Geometrical and topological limitations on the structures of molecules and crystals 20

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### **Structural Inorganic Chemistry : Wells,a.f. : Free ...**

This textbook offers over 400 problems and solutions in structural inorganic chemistry for senior undergraduates and beginning graduates. It is an updated companion text to Advanced Structural Inorganic Chemistry by the same authors. The new edition adds over 100 new problems and three new chapters on metal compounds and bioinorganic chemistry.

### **Inorganic Chemistry - Oxford University Press**

Inorganic chemistry deals with synthesis and behavior of inorganic and organometallic compounds. This field covers all chemical compounds except the myriad of organic compounds (carbon-based compounds, usually containing C-H bonds), which are the subjects of organic chemistry. The distinction between the two disciplines is far from absolute, as there is much overlap in the subdiscipline of ...

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The basic crystal structures introduced include cubic closest packing (NaCl, CaF<sub>2</sub>, cubic ZnS, spinel compounds, etc.), hexagonal closest packing (hexagonal ZnS, NiAs, CdI<sub>2</sub>,  $\alpha$ -Al<sub>2</sub>O<sub>3</sub>, rutile, etc.), body-centered cubic packing ( $\alpha$ -AgI, CsCl, etc.).

### **Basic Inorganic Crystal Structures and Materials - Oxford**

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Models in structural inorganic chemistry by Wells, A. F., 1970, Oxford University Press edition, in English

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